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relationship between the reaction rate and the concentrations of reactants. • The exponents tell the order of the reaction with respect to each reactant. • Since the rate law is  $\text{Rate} = k [\text{NH}_4^+] [\text{NO}_2^-]$  the reaction is first-order in  $[\text{NH}_4^+]$  and first-order in  $[\text{NO}_2^-]$ .

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