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work As shown in
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to a pile of orange
ammonium dichromate
powder results in a
vigorous reaction
known as the
ammonium dichromate
volcano. Heat, light,

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and gas are produced as a large pile of fluffy green chromium(III) oxide forms.

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A representation of a chemical reaction with reactants on the left, products on the right, and an arrow separating the two. Small whole number ratios that are placed

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in front of the formulas
in an equation in order
to balance it.

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A combustion reaction is any reaction that burns in the presence in oxygen gas (O_2).

The products are usually an oxide of the element. If there is the combustion of a hydrocarbon, carbon dioxide and water

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vapor are products. CH
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11.3.1, applying a
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to a pile of orange
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powder results in a
vigorous reaction
known as the

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ammonium dichromate volcano. Heat, light, and gas are produced as a large pile of fluffy green chromium(III) oxide forms. We can describe this reaction with a chemical equation An expression that gives the identities and quantities of the ...

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Types of Chemical

Reactions Often

charcoal briquettes

provide the heat for

barbeque grills through

the burning of carbon.

Have you ever felt the

heat and smelled the

smoke coming from a

burning charcoal grill?

The heat and smoke

are the products of a

combustion reaction.

Combustion is one of

the five general types

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set (253) chemical

equation _____ is

a representation of a

chemical reaction with

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reactants on the left,
products on the right,
and an arrow
separating the two.

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The process of breaking food eaten down in the body in order to absorb what is needed and more easily eliminate - this involves chemical reactions. Define:
Chemical Equation: A "chemical equation" is a representation of a chemical reaction; the formulas of the "reactants" are connected by an "arrow (->)" with the formulas of the

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Chemical
"product".
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